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Development and Optimization of Multi-Functional SCR-DPF Aftertreatment for Heavy-Duty NOx and Soot Emission Reduction

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2018 Annual Merit Review, DOE Vehicle Technologies Office Arlington, VA June 20, 2018

Project ID: ACS119

Overview



Timeline

- 4-yr CRADA
 - Start date July 2016
 - End date June 2020
- 41.7% complete

Budget

- Contract value \$2.7M
 - \$1.35M DOE share
 - \$1.35M PACCAR share
- Funding received
 - FY16 \$200K
 - FY17 \$355K
 - FY18 \$175K

Barriers

- B. Lack of cost-effective emission control for meeting EPA standards for NOx & PM
- **E. Durability** of the emission control system: 435,000 miles
- ► **G. Cost** of emission control devices ... for heavy duty engines in particular

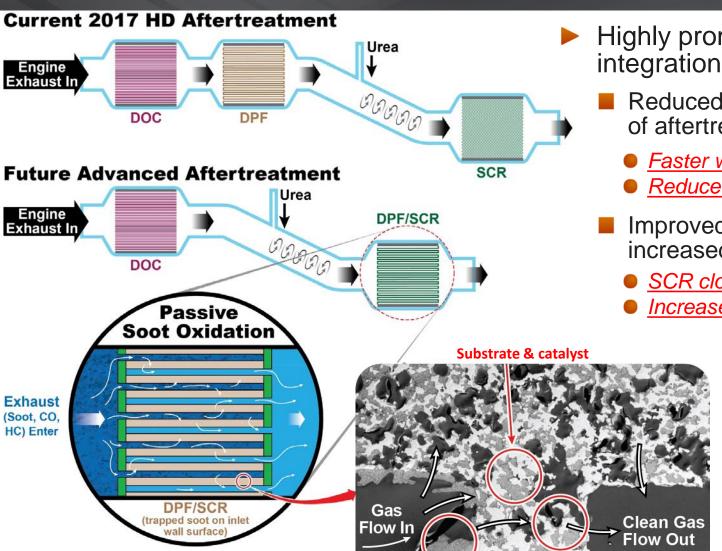
Partners



CRADA partner

Relevance Multi-Functional Aftertreatment: SCR-on-DPF





Highly promising strategy of

- Reduced total thermal mass of aftertreatment system
 - Faster warm up
 - Reduced cold start emissions
- Improved performance & increased flexibility
 - SCR closely coupled to engine
 - Increased SCR volume

Soot trapped upstream

Molecular diffusion to washcoat

Relevance SCR-on-DPF (for heavy duty application)



- Challenges to deployment
 - 1. SCR catalyst performance light/heavy duty
 - Enablers ultra-high porosity filter development, advanced imaging techniques
 - 2. SCR catalyst durability light/heavy duty
 - ☐ Enablers small-pore Cu-zeolites, e.g., Cu/SSZ-13. more thermally durable
 - 3. Passive soot oxidation performance (via NO₂) heavy duty
 - □ Economically attractive to manage soot passively for heavy duty
 - With incorporation of SCR phase, <u>competition for NO₂</u>

Fast-SCR
$$2NH_3 + NO + NO_2 \rightarrow 2N_2 + 3H_2O$$

Dominates NO_2 consumption versus

Significantly compromises soot oxidation oxidation

 $C (soot) + 2NO_2 \rightarrow co_2 + 2NO$

ApproachModify the SCR catalyst to generate NO₂ in situ



- Focus Incorporating <u>increased NO oxidation (to NO₂)</u> through inclusion of a metal-oxide component for selective catalytic oxidation (SCO)
 - i.e. forming an SCR-SCO binary catalyst system (as a single active phase)

PROJECT OBJECTIVES

- NO oxidation
 - Target: increased NO oxidation to NO₂
 - … that then drives increased passive soot oxidation
- SCR catalyst performance [i.e., NOx reduction efficiency (NRE)]

Must develop in *integrated* fashion

- Target: NO oxidation without oxidizing NH₃
- Target: NRE ≥ current state-of-the-art (Cu-CHA)
- SCR catalyst durability
 - Target: durability ≥ current state-of-the-art (Cu-CHA)

Approach Modify the SCR catalyst to generate NO₂ in situ



Focus – Incorporating <u>increased NO oxidation (to NO₂)</u> through inclusion of a metal-oxide component for selective catalytic oxidation (SCO)

PROJECT OBJECTIVES this past year

- NO oxidation PACCAR focus
- SCR catalyst performance and durability PNNL focus

■ Step 1 – SCO substrate

- SCO is envisioned as a ZrO₂-based catalyst
- Understanding the impact of ZrO₂ in close proximity to the SCR catalyst
- including the impact on both SCR performance and durability

Step 2 – SCO catalyst

- Impact of hetero-atoms included with ZrO₂
 - Inside the lattice structure, outside the lattice structure, etc.
- Including their impact on SCR performance and durability

Approach Schedule and Milestones

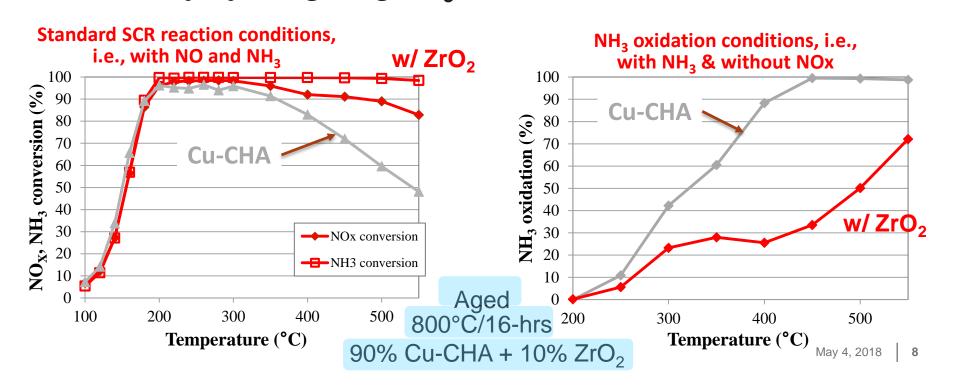


Date*	Milestone and Go/No-Go Decisions	Status
November 2017	Milestone: 1st group of optimized SCRF samples with candidate SCO/SCR binary catalyst ready for detailed testing	Complete
February 2018	Go/No-Go decision: Identify candidate SCO/SCR binary phase catalyst with improved soot oxidation performance with competing SCR	Delayed
February 2018	Milestone: Composition for first increased scale testing	Complete (PACCAR)
May 2018	Milestone: 2 nd group of optimized SCRF samples with candidate SCO/SCR binary catalyst ready for detailed testing	On-track
August 2018	Milestone: SCRF single wall model complete (without aging behavior)	In pursuit
August 2018	Milestone: Production of catalyst for first increased scale testing	Complete

Accomplishments SCR catalyst performance and durability



- ZrO₂-based metal-oxide SCO catalyst
 - Focus Understanding the impact of ZrO₂ on SCR performance and durability
 - Baseline Cu-CHA (Si/Al = 12, Cu ~2.3 wt%, IE level ~30%)
- ZrO₂ addition leads to significantly improved high temperature selectivity by mitigating NH₃ oxidation





Accomplishments SCR catalyst performance and durability



Cu-CHA + 10% ZrO₂

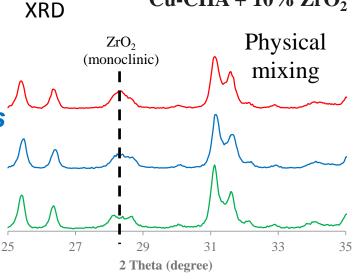
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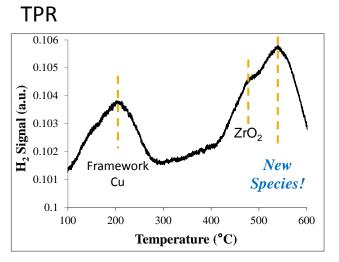
ZrO₂-based metal-oxide SCO catalyst

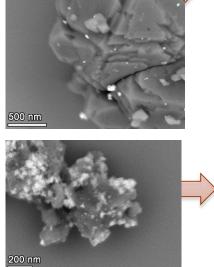
Focus – Understanding the impact of ZrO₂

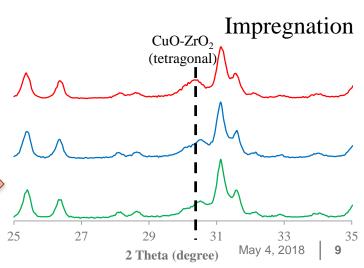
► Identified the generation of a new species resulting from the interaction between extra-framework Cu (i.e., CuO) and ZrO₂

Closer vicinity = Increased interaction between extra-framework Cu and ZrO₂







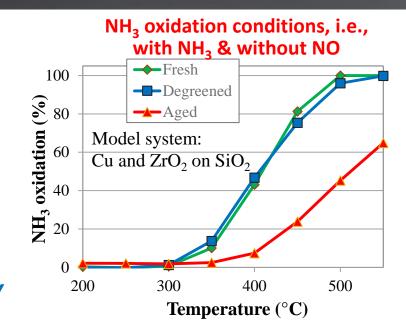


Accomplishments SCR catalyst performance and durability



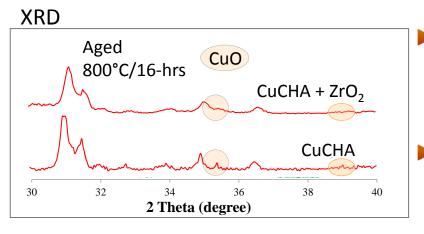
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- ZrO₂-based metal-oxide SCO catalyst
- ZrO₂ addition and extra-framework Cuincorporation as hetero-atom
- Demonstrates the ability to incorporate hetero-atoms into the ZrO2 matrix (e.g., for improving NO oxidation) while mitigating NH₃ oxidation and the adverse impact on SCR performance, and improved durability



i.e. the worm

in apple



 $CuO_x \rightarrow inferior$ high-T selectivity

and

 $CuO_x \rightarrow damages$ the zeolite



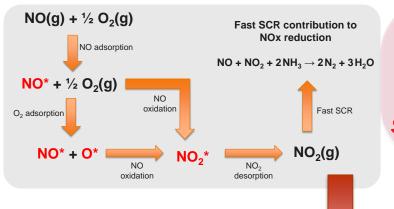
Song, et al., ACS Catal., 7-12 (2017) 8214-8227 May 4, 2018

Accomplishments Taking advantage of a NOx storage component



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► Including Ba – 90% Cu-CHA + 10% Ba/Zr-oxide



Standard SCR

reaction conditions

180

190

Increased

Ba

Temperature (°C)

160

95

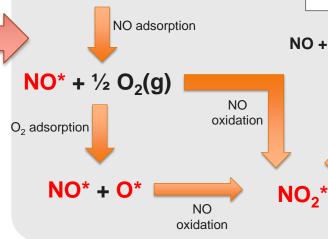
90

150

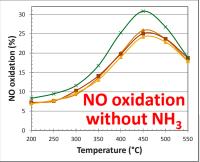
Conversion (%)

Ba/Zr-oxide shows evidence of the ability to take advantage of a surface-active species

 $NO(g) + \frac{1}{2}O_2(g)$



- With Ba, resolving ...
 - Analogous NO oxidation activity
 - Increased NOx reduction efficiency



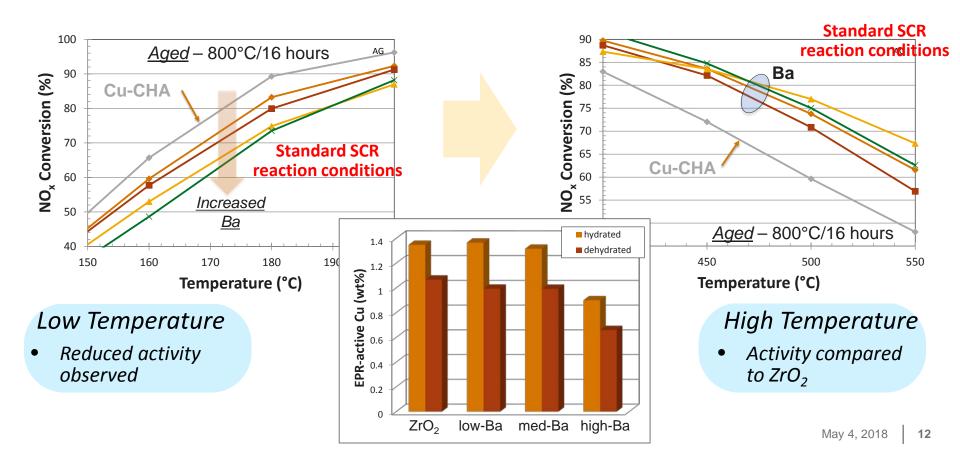




Accomplishments Implications of Ba on SCR catalyst durability



- ► Including Ba 90% Cu-CHA + 10% Ba/Zr-oxide
- Work has uncovered an aging mechanism that uniquely affects low-temperature activity of BaO/ZrO₂ versus Cu-CHA or with ZrO₂



Accomplishments SCR catalyst durability



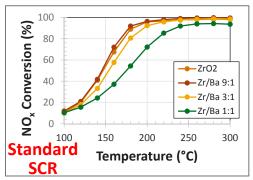
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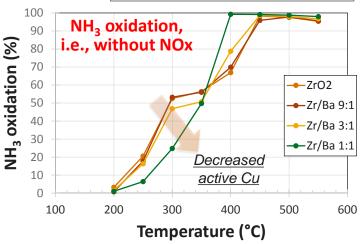
- ► Including Ba 90% Cu-CHA + 10% Ba/Zr-oxide
- Thermally-induced ion-exchange aging mechanism
 - = function (time, temperature, proximity)

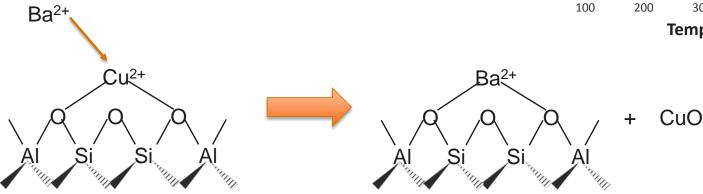
Ion-exchange aging behavior ...

- > 800°C/16-hrs >> 650°C/100-hrs
- Closer vicinity >> physical mixture

Must be cognizant of aging mechanism in relation to SCO catalyst chemistry AND proximity to SCR phase







Responses to Previous Years Reviewers' Comments

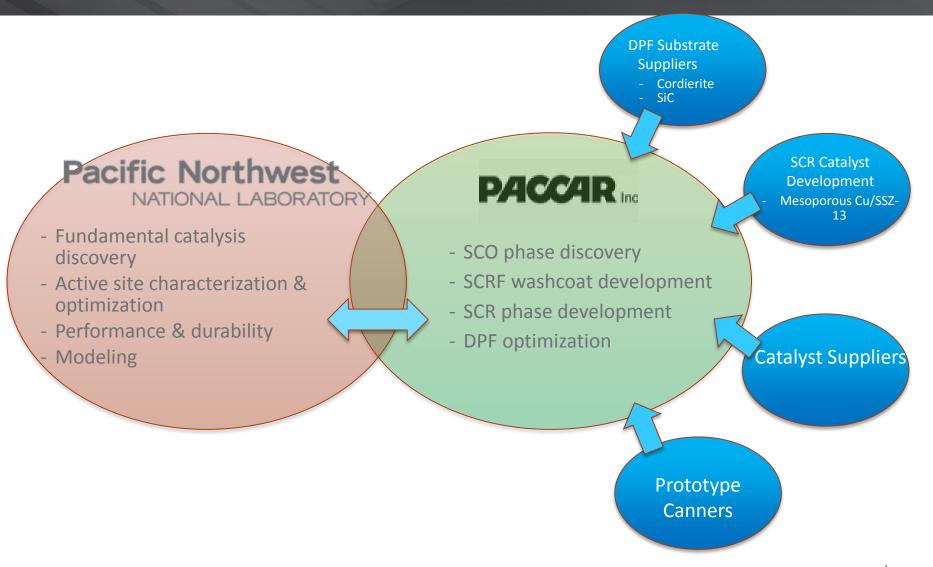


- ► The reviewer opined that work on ZSM-5 is a waste of time. This reviewer affirmed that moving to SSZ-13 is correct. The reviewer further remarked that the most important part of the project is the durability study, and S effects need to be included.
 - We agree that durability must be intimately considered in the development. Thus, this has been PNNL emphasis in the past year
- ► This reviewer commented that the approach seems naive. This reviewer stated that it will be extremely difficult, if not impossible, to balance high SCR conversion with high soot oxidation.
 - With current state-of-the-art SCR technology, we do not disagree. Thus, we are trying modify current SCR technology to enable the application. This is the novelty of the work.
- The reviewer said that work with Cu-ZSM-5 was not relevant. The reviewer noted that it is unclear how the filters are being coated.
 - ZSM-5 was employed solely as a model system, and we have moved past this stage.
 - Filters are being coated by PACCAR



Collaboration and Coordination

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Remaining Challenges and Barriers

- NO oxidation
 - Increased NO oxidation to NO₂
 - … to drive increased passive soot oxidation
- ► NOx reduction efficiency (NRE)
 - Increasing NO oxidation without oxidizing NH₃

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Proposed Future Work

NO oxidation

- Modifying ZrO₂ chemistry (increased redox, acidity)
- More aggressive active NO oxidation component (e.g., Mn, Co, ...)
- ... with and without Ba

NOx reduction efficiency (NRE)

- Incorporating (Mn, Co, ...) into ZrO₂-based lattice
- Prior work metals in a support lattice structure (i.e., co-precipitated) can be just as active for NO oxidation versus supported
- Leveraging this with the ability of ZrO₂ to mitigate NH₃ oxidation activity of hetero-atoms (as shown from extra-framework Cu interaction)

Any proposed future work is subject to change based on funding levels

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Summary

- ZrO₂ addition shown to improve performance by interacting with extraframework Cu (i.e., CuO) and ZrO₂
 - Generation of a new species resulting from the interaction
 - Significantly improved high temperature selectivity by mitigating NH₃ oxidation
 - Closer vicinity = Increased interaction between extra-framework Cu and ZrO₂
- ZrO₂ demonstrates the ability to incorporate hetero-atoms into matrix (e.g., for improving NO oxidation) while mitigating detrimental SCR impact
- Ba/Zr-oxide shows evidence of the ability to take advantage of a surfaceactive species as a pathway to enhanced NO oxidation behavior
- Uncovered an ion-exchange (IE) aging mechanism that uniquely affects low-temperature activity of BaO/ZrO₂ versus Cu-CHA or with ZrO₂
- Development must be cognizant of IE aging mechanism in relation to SCO catalyst chemistry AND proximity to SCR phase



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Technical Backup Slides

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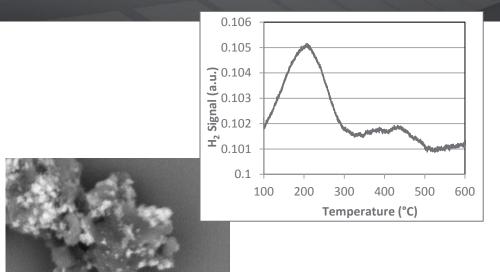
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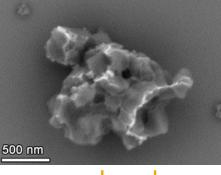


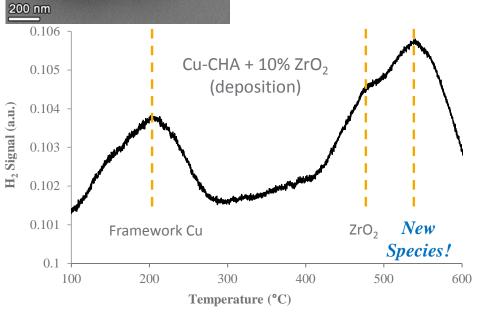
Backup – Enhancing the Cu/ZrO₂ interaction

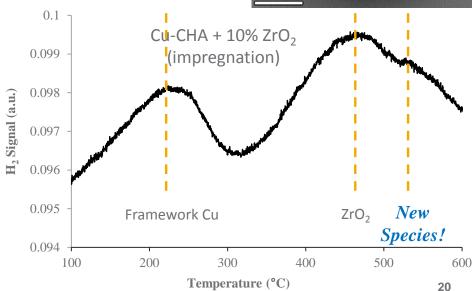




New species observed in the TPR profile of deposited- and impregnated-ZrO₂ on Cu-CHA



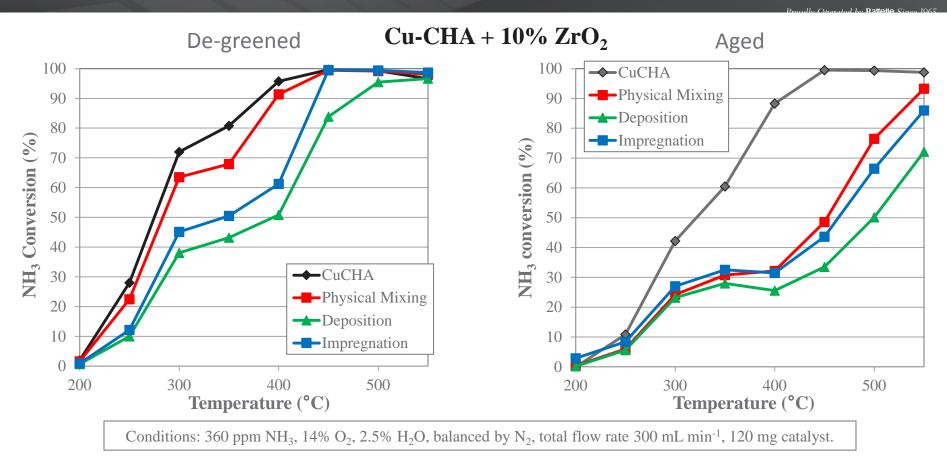






Backup – Reduced NH₃ oxidation (by O₂)



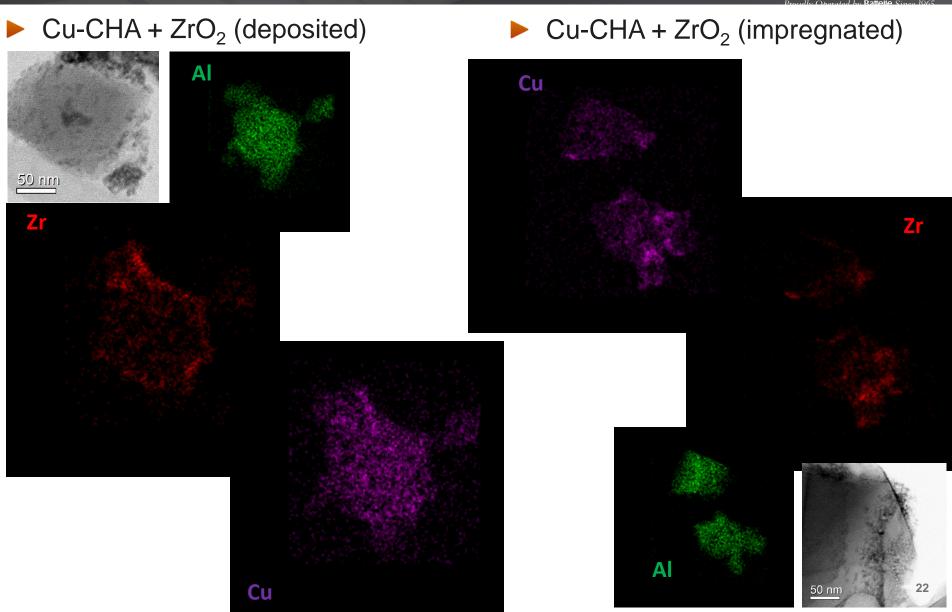


- Extra-framework Cu is predominantly responsible for the high temperature (400° C+) NH₃ oxidation activity.
- ► NH₃ oxidation directly affected by the generation of a new species resulting from the interaction between extra-framework Cu and ZrO₂.



Backup – (S)TEM-EDS observed Cuenrichment of ZrO₂



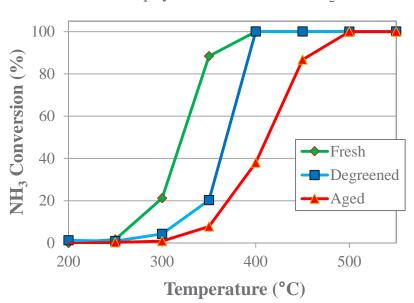


Backup – Model system study

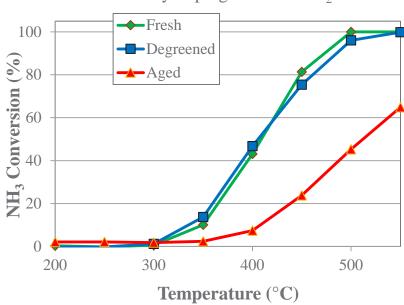


Intent of model catalyst is to mimic the interaction between extra-framework Cu (responsible for enhanced high temperature NH₃ oxidation) and ZrO₂. Inert SiO₂ is used as an inert substitute for Cu-CHA.

Catalyst 1: Cu impregnated on ZrO₂, then physical mixed with SiO₂



Catalyst 2: Cu and ZrO₂ simultaneously impregnated on SiO₂



Conditions: 360 ppm NH₃, 14% O_2 , 2.5% H_2O , balanced by N_2 , total flow rate 300 mL min⁻¹, 120 mg catalyst.

A close vicinity between Cu and ZrO₂ is necessary for reduced NH₃ oxidation activity and inducing the interaction between extra-framework Cu and ZrO₂.

Backup – Understanding BaO/ZrO₂ aging



- Aged 800°C/16 hours
- Comparing to aging at 650°C/100-hrs
 - Significantly different at low temperature
 - Very similar at high temperature

Confirms aging mechanism is unique, and NOT simply a propagation of conventional Cu-CHA aging



